

US EPA ARCHIVE DOCUMENT

# *Risks Posed by Brines Containing Dissolved CO<sub>2</sub>*

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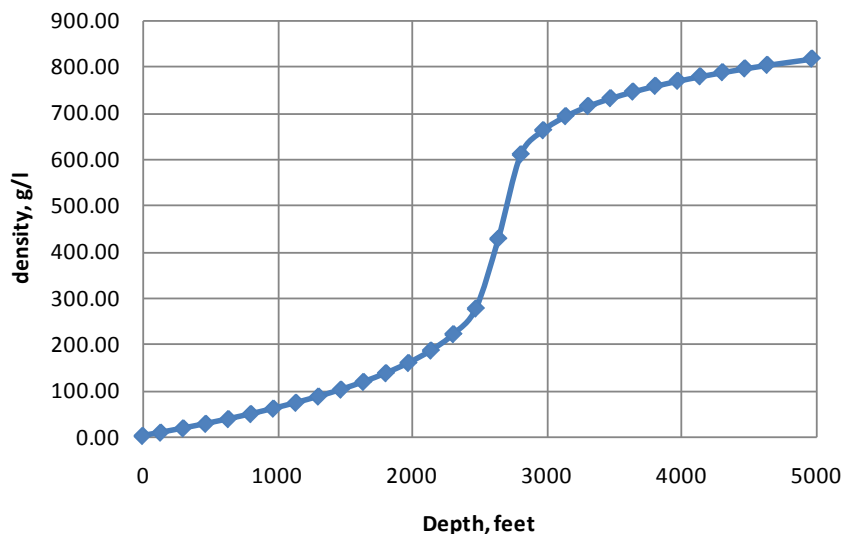
*<sup>2</sup>Stanford University*

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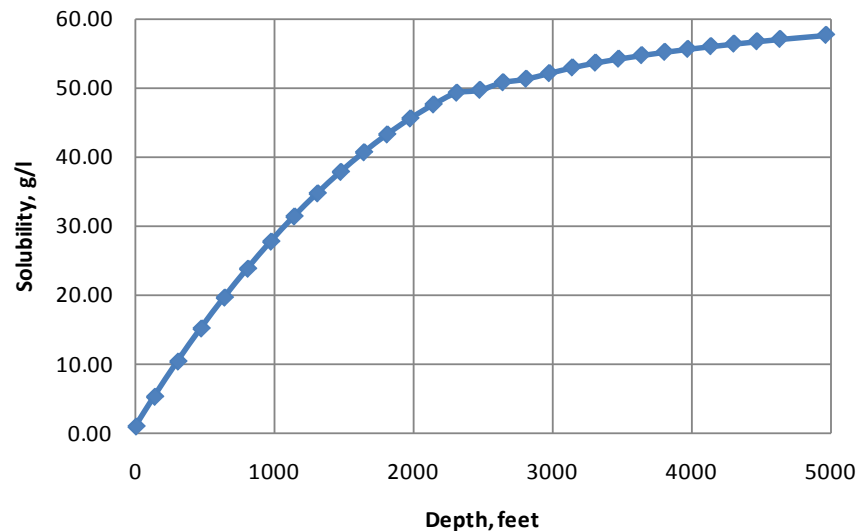


# CO<sub>2</sub> Density and Solubility with Depth

## CO2 Phase Density



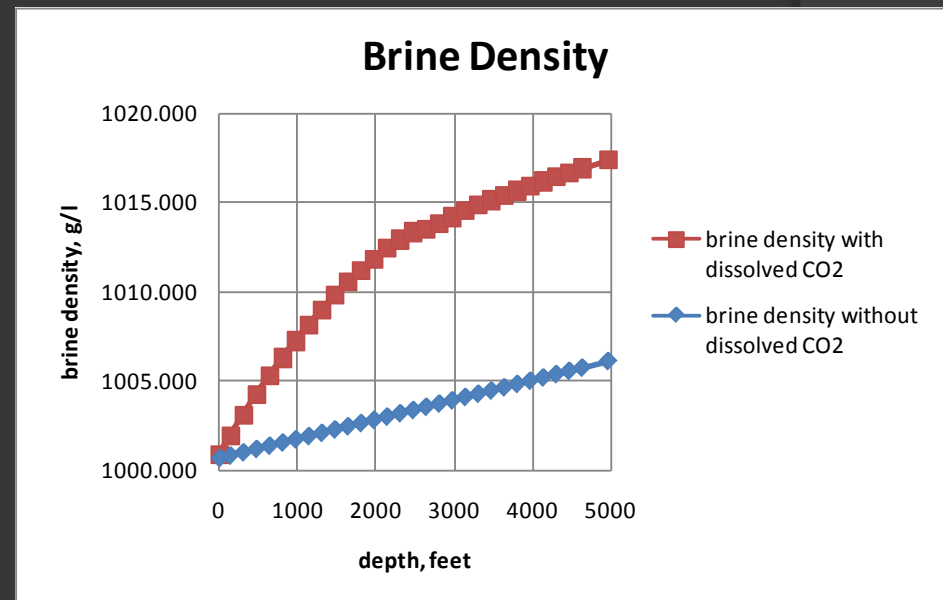
## CO2 Solubility



Calculated using TOUGH2-ECO2N assuming 35° C and 10,000 mg/l NaCl

# The high CO<sub>2</sub> solubility is significant

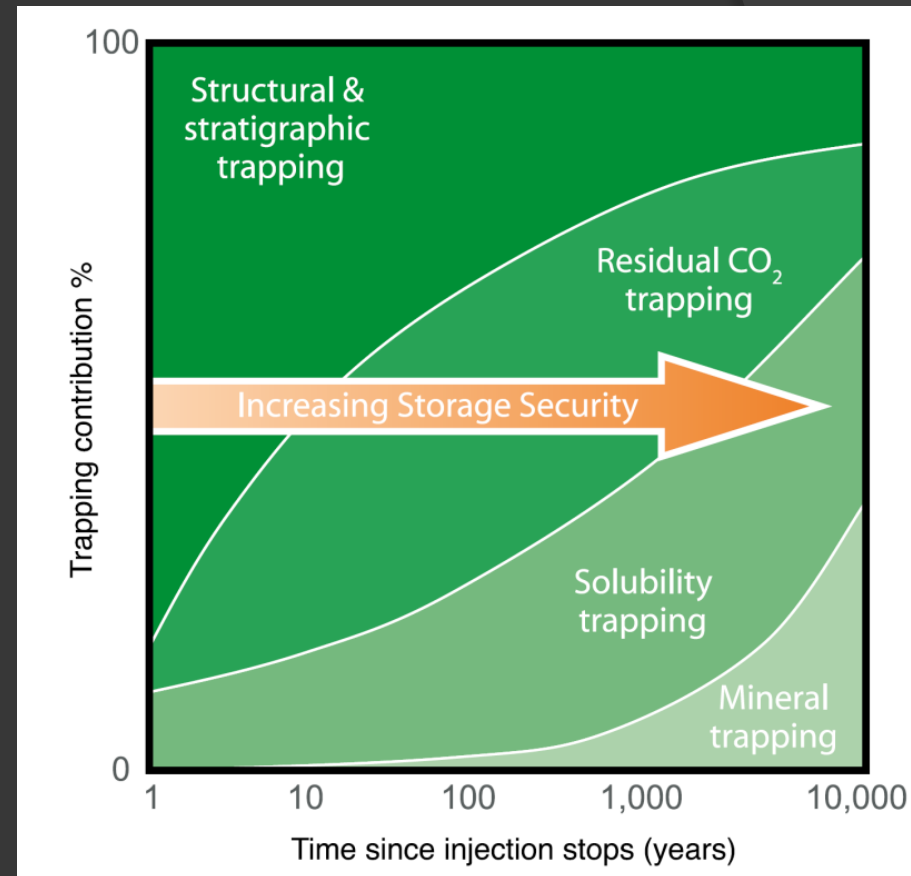
- At 3000 ft depth, we get ~50 g/l (50 times more CO<sub>2</sub> than beer!)
- When CO<sub>2</sub> dissolves, the aqueous phase becomes more dense (about 1% here)
- Upward flow would require a caprock defect, and an upward hydraulic gradient > density difference



Calculated using TOUGH2-ECO2N

# The Dissolved CO<sub>2</sub> is Secure – Or Is It?

- Solubility trapping – CO<sub>2</sub> dissolves in pore water (up to 60 g/l)
- Density increase favors downward flow of CO<sub>2</sub> saturated brine
- Upward flow would require a caprock defect, and an upward hydraulic gradient > 1%
- **However, if a CO<sub>2</sub> saturated brine moved upward, the CO<sub>2</sub> would come out of solution (exsolve), leading to a potentially mobile gas phase**



IPCC, 2005

# Outline

## ⊙ Experiments

- Pore
- Core
- Relative permeability

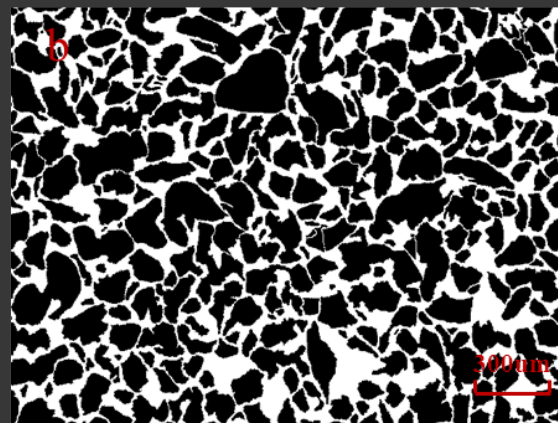
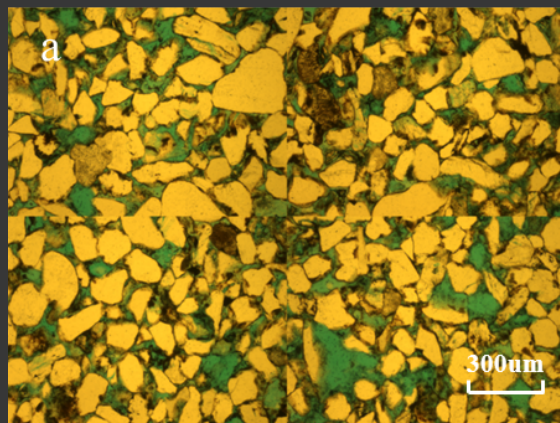
## ⊙ Modeling

- Fault
- Wells
- Dissolved and supercritical injection
- Outcrop

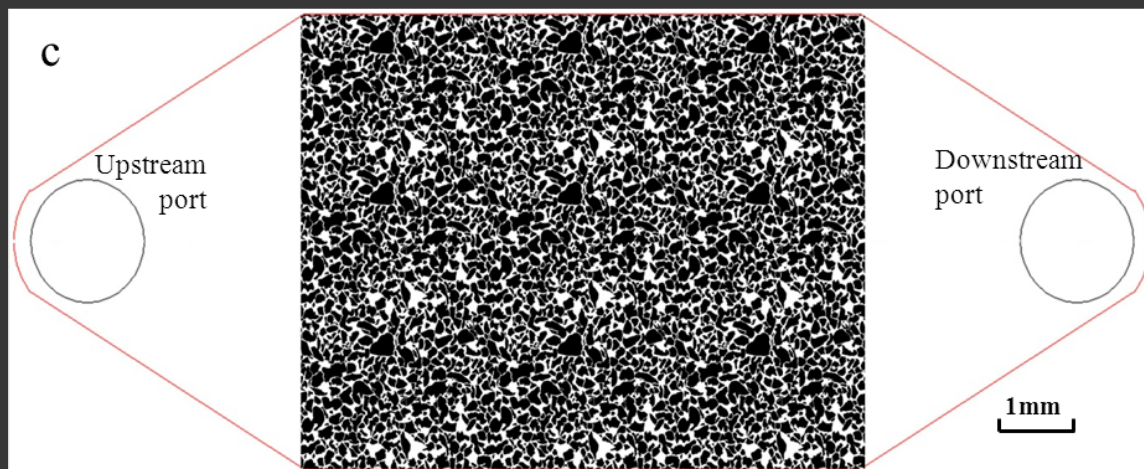
# Laboratory Micromodel Study

(Zuo, Zhang, Falta, and Benson, AWR, 2013)

Thin section micrograph of Mt. Simon sandstone



Binary image used for micromodel

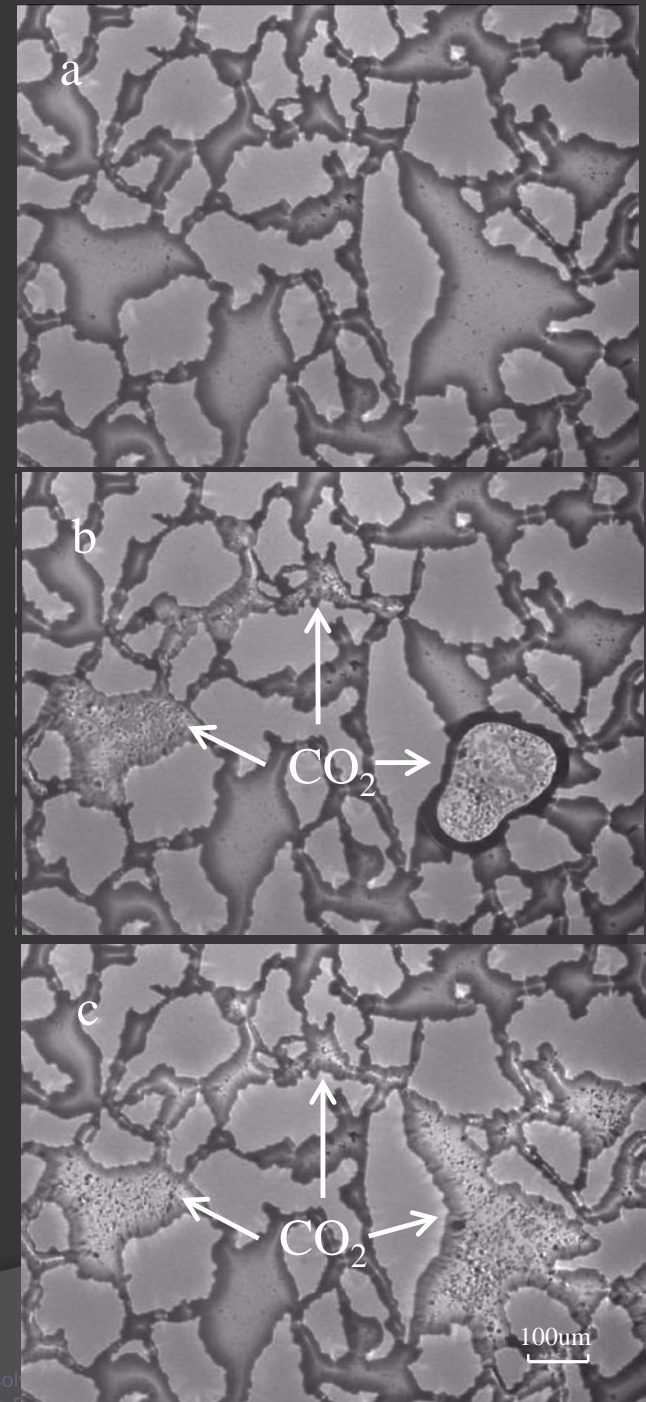


Micromodel: 530 mD; PV=1.35 uL



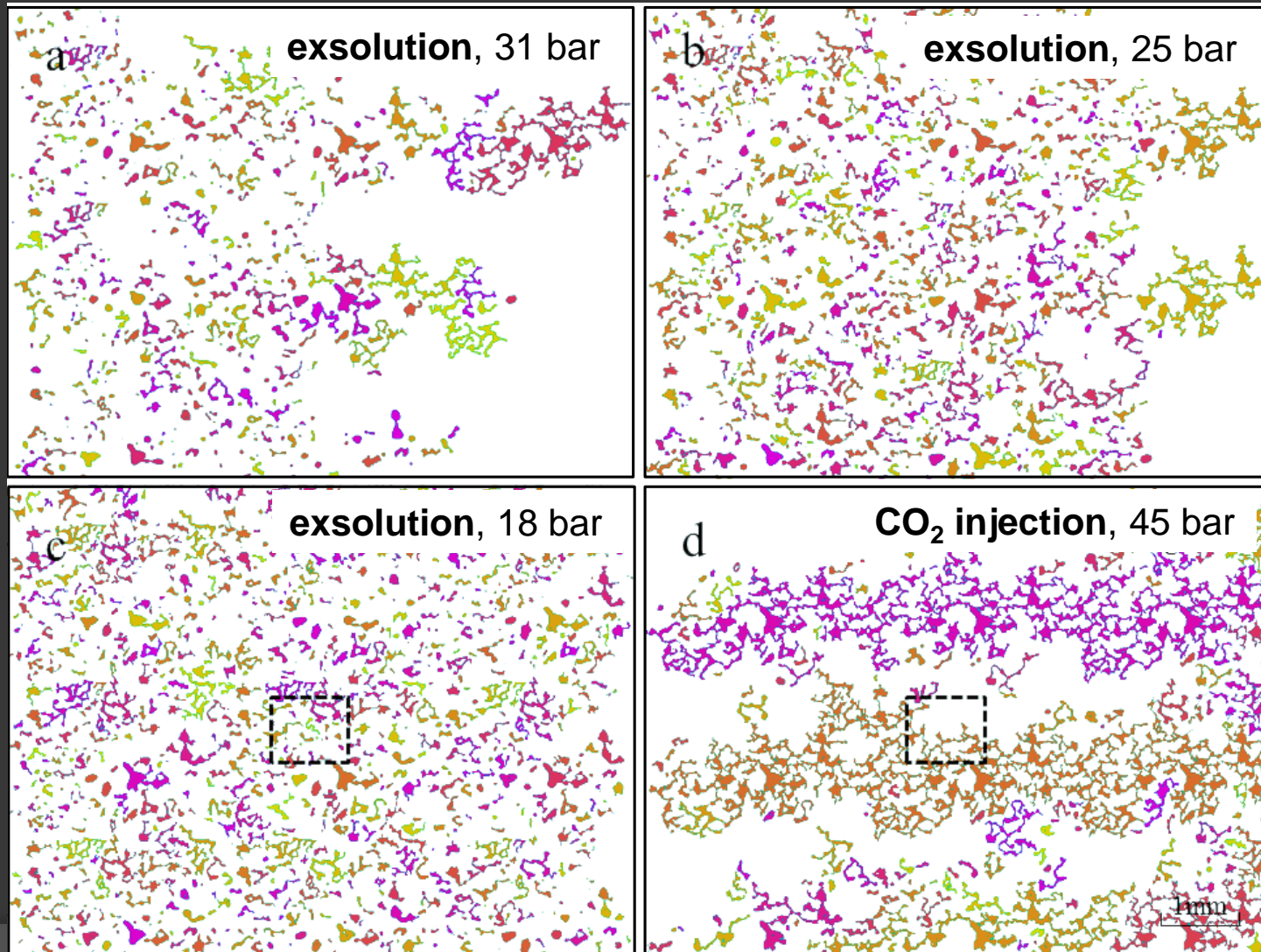
# Micromodel

- Initially fill micromodel with water saturated with dissolved  $\text{CO}_2$  at 90 bars, 45 °C
- Depressurize at a rate of 10 bars/hr
- Images taken at 1 second intervals after onset of exsolution at 31 bars
- $\text{CO}_2$  first starts to flow out at 23.5 bars, with a  $\text{CO}_2$  phase saturation of 56%



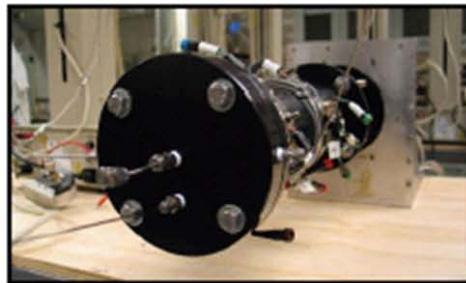


# Comparison of Exsolution and Supercritical CO<sub>2</sub> Injection

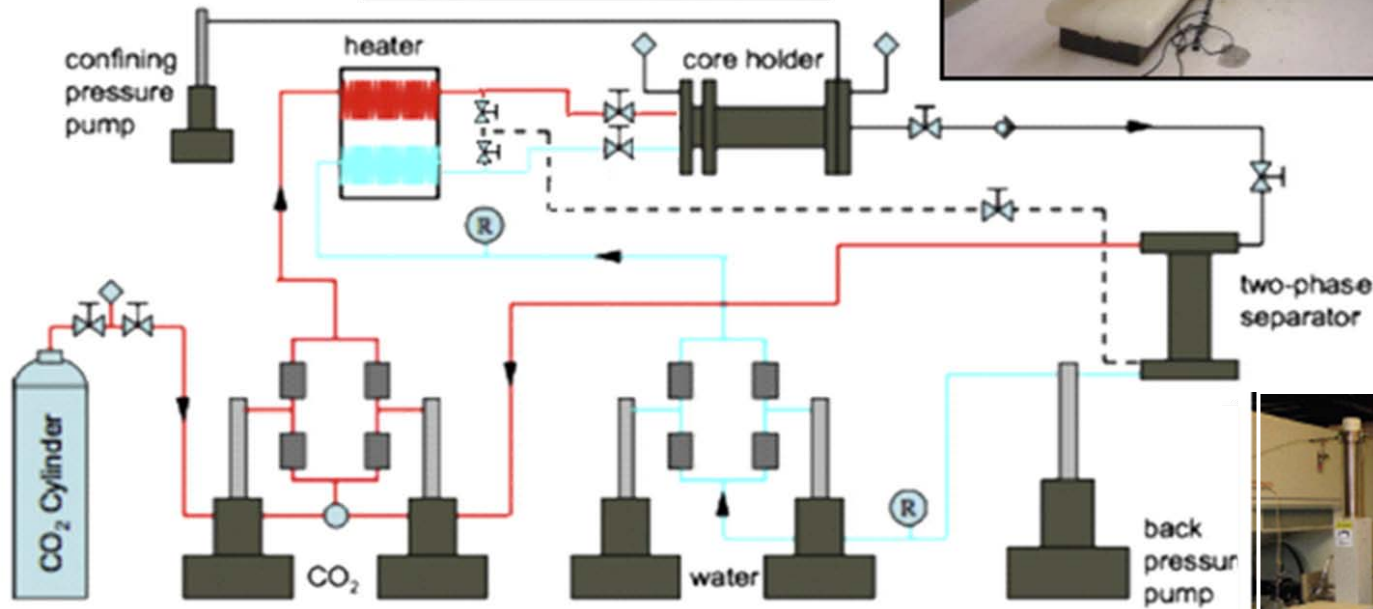


# Core Scale Experimental Setup

**Core Holder**



**CT Scanner**



**Dual-pump System**

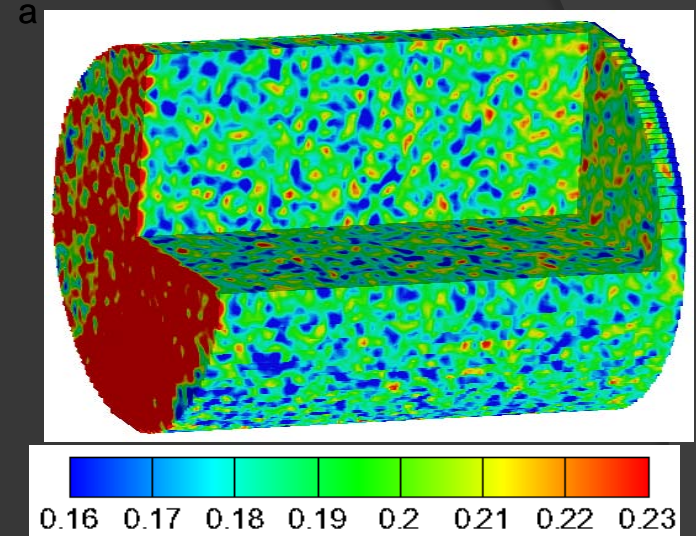
# Core Experiments

## Mobility of exsolved gas

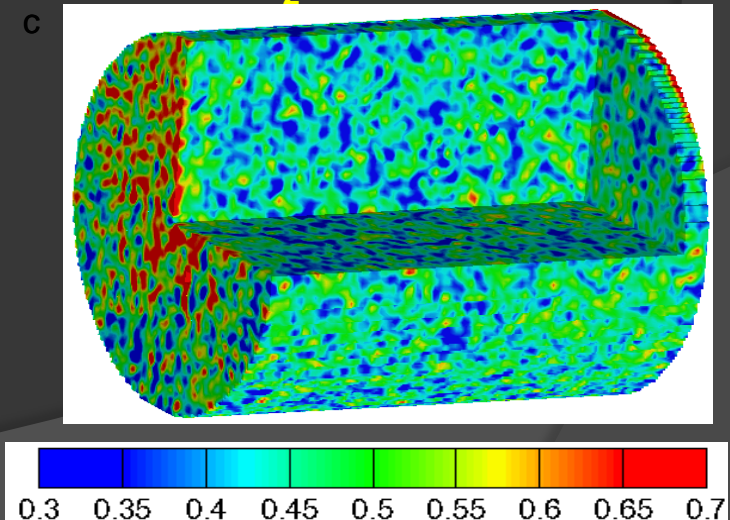
(Zuo, Krevor, Falta, and Benson, TIMP, 2012)

- Fill core with CO<sub>2</sub> saturated water at 124 bar, 50 °C
- Depressurize to 27 bar at a rate of 12 bars/hr
- CO<sub>2</sub> phase saturation reaches >40%, but very low mobility
- No gravity redistribution after 11 days.
- CO<sub>2</sub> is mobile at 3% gas saturation during flood of the same core

porosity



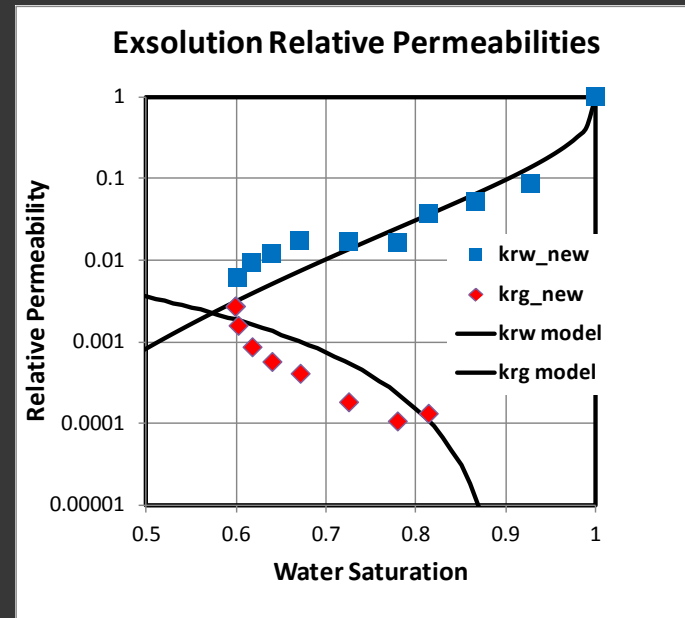
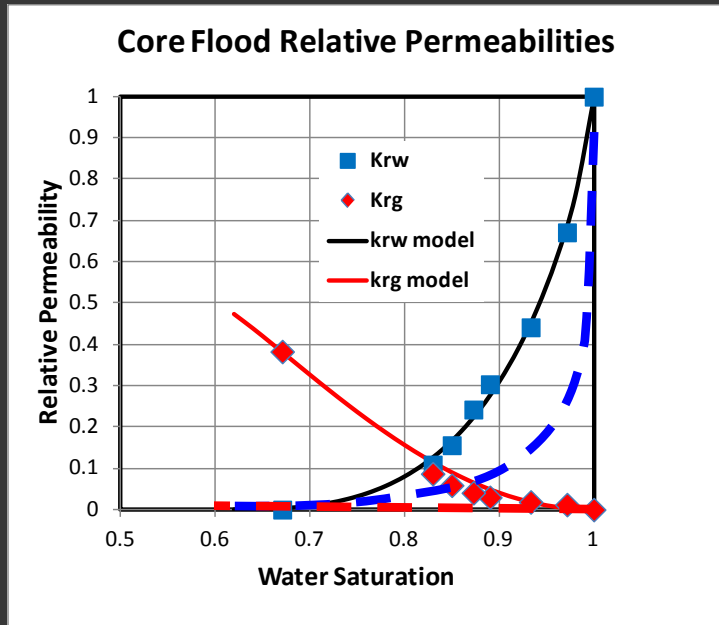
CO<sub>2</sub> saturation



# Core Experiments

## Relative permeability

Mt. Simon Sandstone (15.7 mD, 23.9 % porosity)



CO<sub>2</sub> phase injection



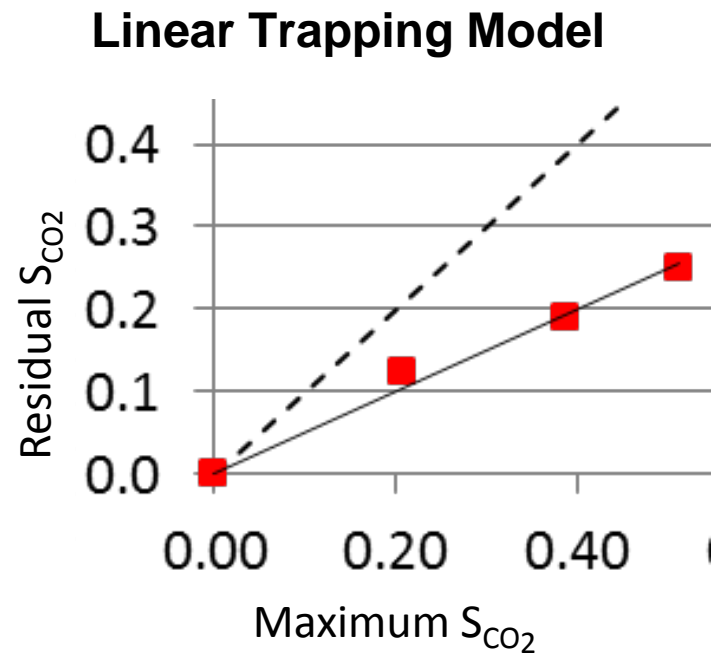
CO<sub>2</sub> exsolution from brine



# Core Experiments

## Hysteretic $\text{CO}_2$ phase trapping

- Core flood experiments where  $\text{CO}_2$  saturation was cyclically increased and decreased to measure trapping
- $\text{CO}_2$  saturation was measured by CT scan
- Trapped  $\text{CO}_2$  is a linear function of maximum  $\text{CO}_2$  saturation



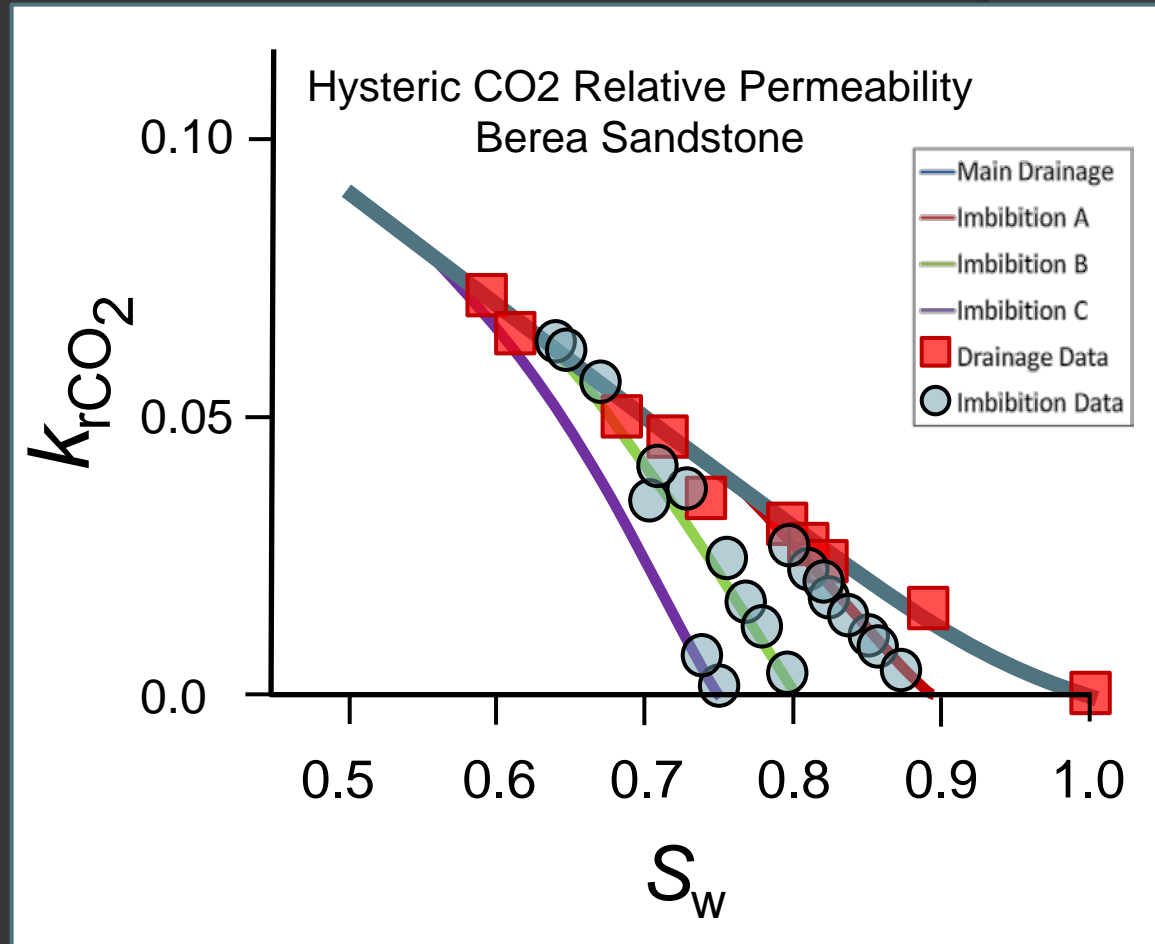


# Core Experiments

New relative perm model for hysteretic CO<sub>2</sub> phase trapping

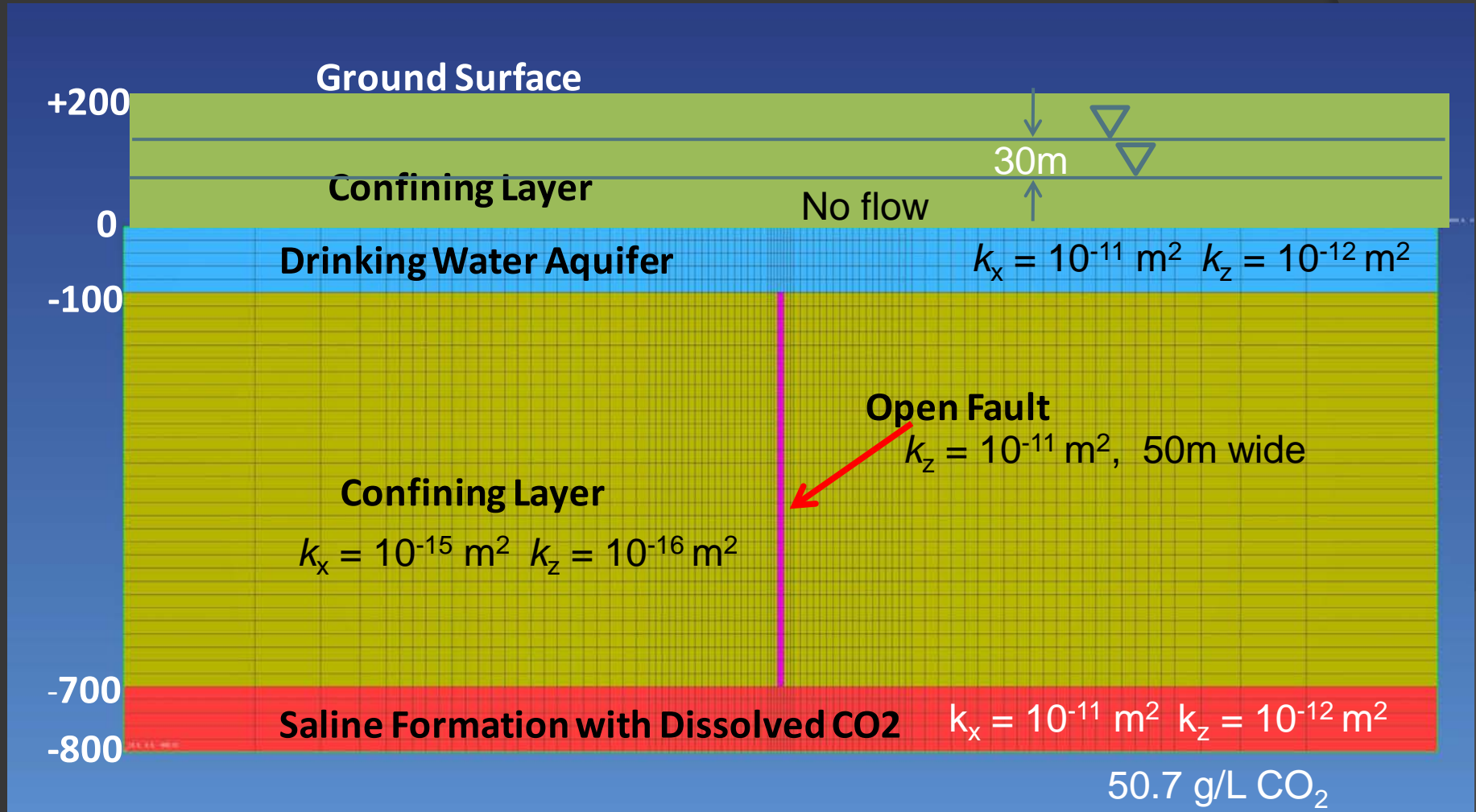
- Simple approach: residual saturation a function of maximum saturation
- Continuously update the max residual saturation
- Allows use of existing relative permeability models

$$k_{rg} = k_{rg \max} \sqrt{1 - \hat{S}_w} \left(1 - \hat{S}_w^{1/m}\right)^{2m}$$
$$\hat{S}_w = \frac{S_w - S_{wr}}{1 - S_{wr} - S_{gr}}$$



# Modeling

Open fault model using TOUGH2-ECO2N

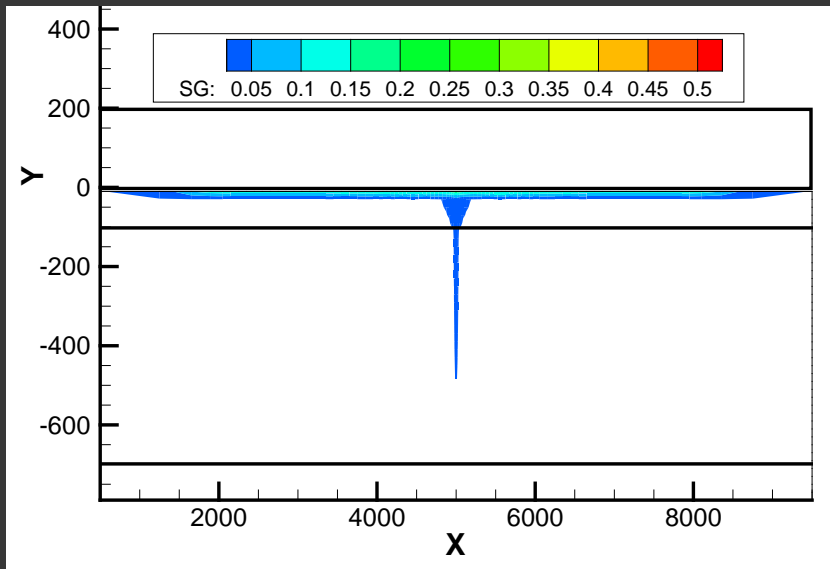


← **Model extends 5,000m in either direction** →

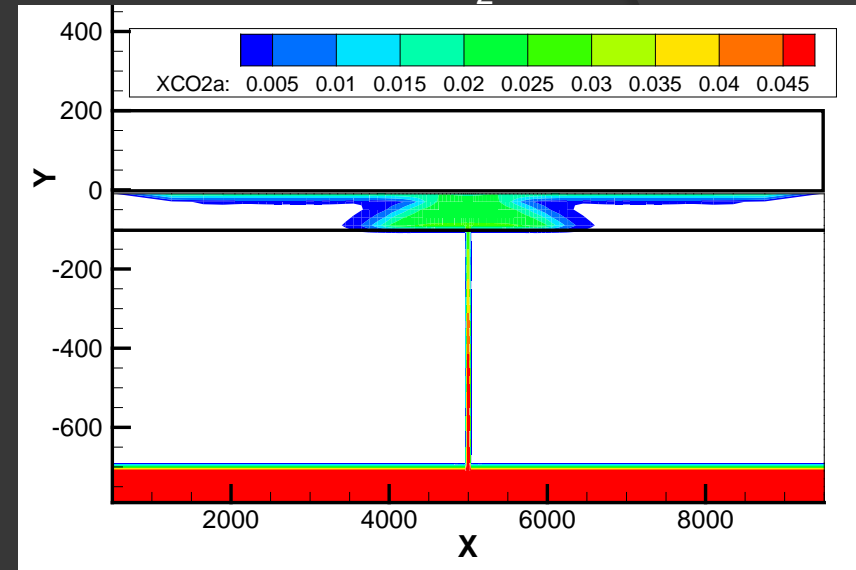


# Model using regular core flood relative permeabilities. Time is 30 years.

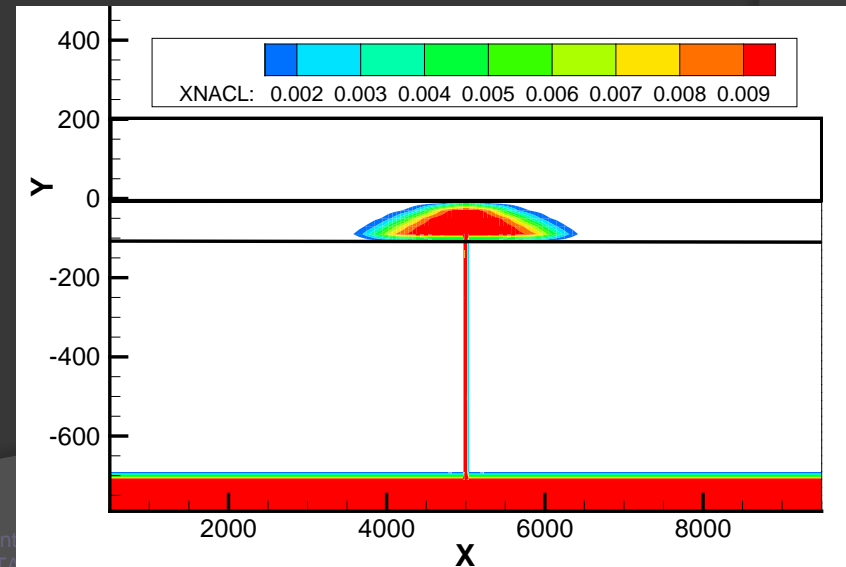
## Gas saturation



## Dissolved CO<sub>2</sub> mass fraction



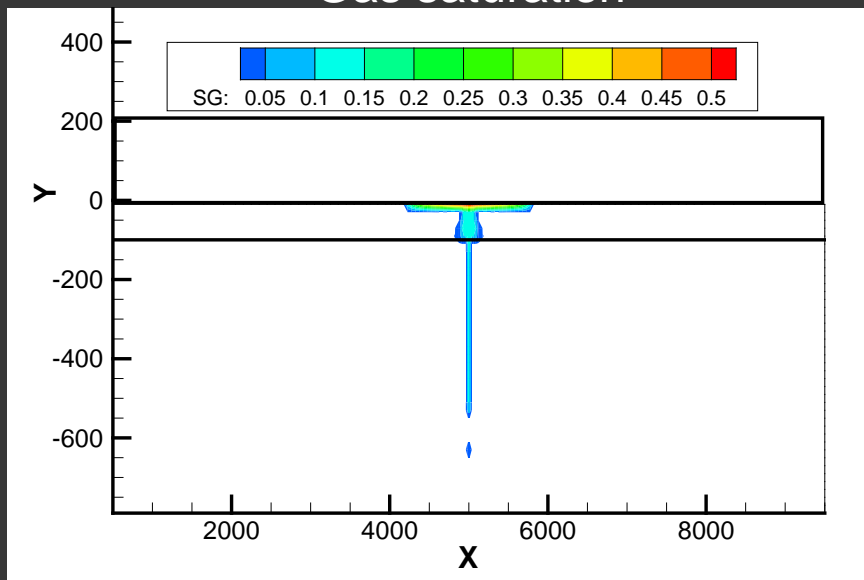
## Dissolved salt mass fraction



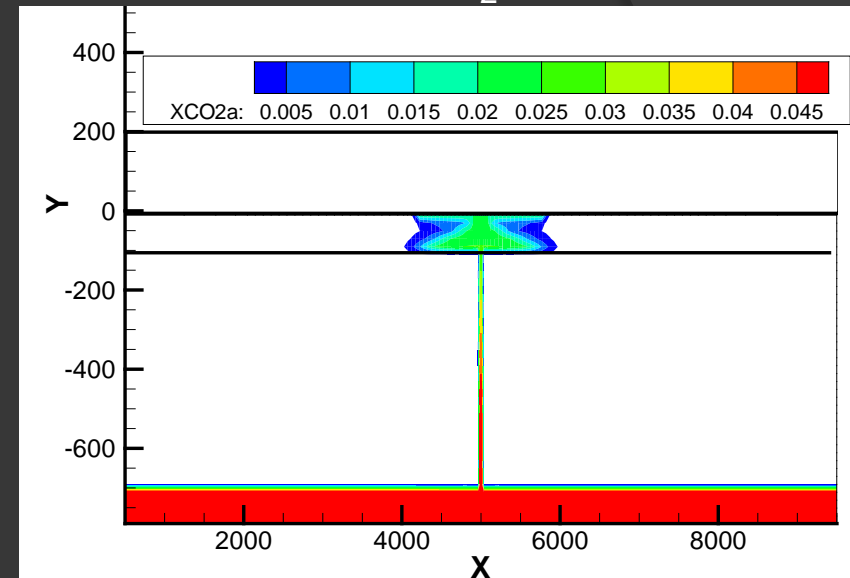
Gas phase CO<sub>2</sub> reaches the DWA, and spreads to the boundaries at 5000m within 30 years if the drawdown is maintained.

# Model using exsolution relative permeabilities. Time is 30 years.

## Gas saturation

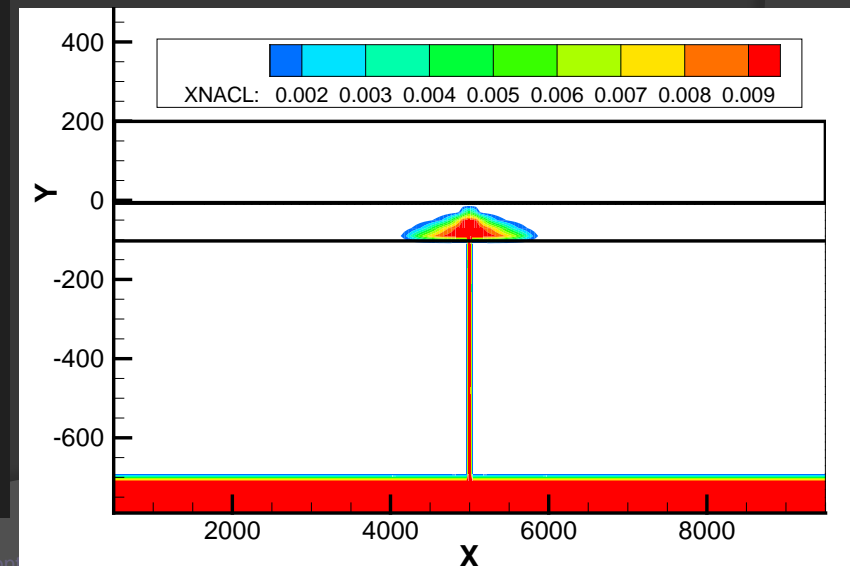


## Dissolved CO<sub>2</sub> mass fraction



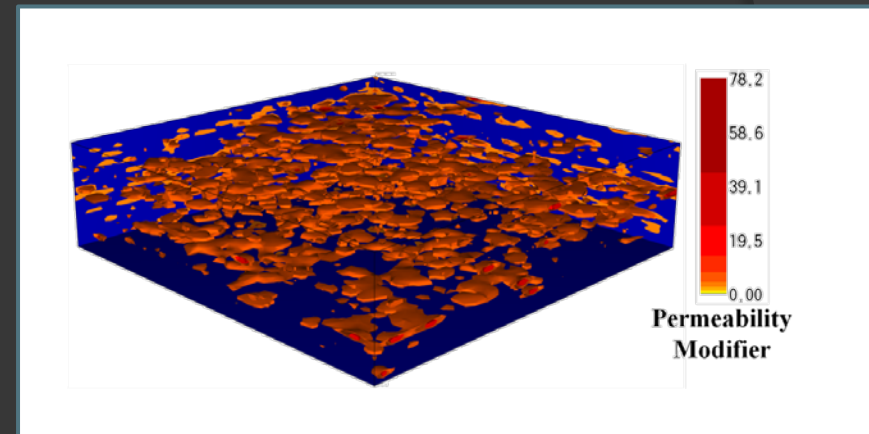
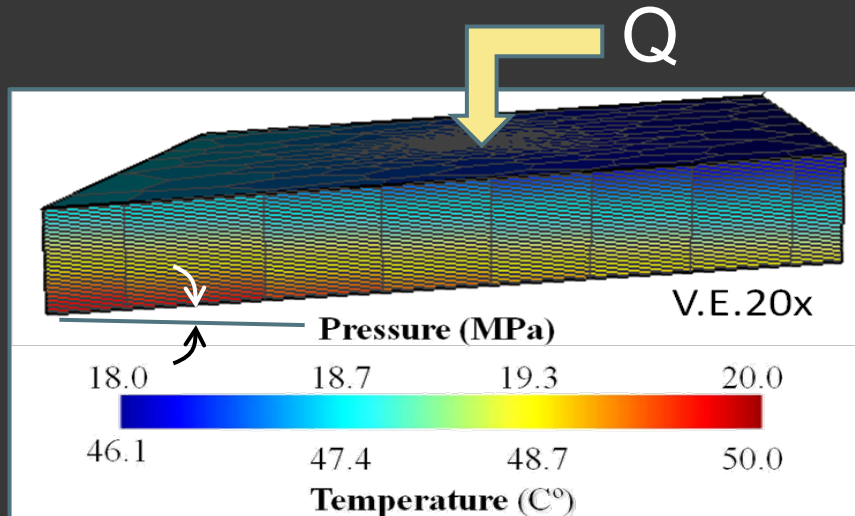
- Leakage much less using exsolution relative permeability
- Related simulations for wells similar
- In all cases, CO<sub>2</sub> migration stops when head imbalance is corrected, no runaway effect

## Dissolved salt mass fraction



# Modeling

## CO<sub>2</sub> injection as dissolved or supercritical



Formation: 300m thick, 20km x 20 km  
Slope: 0.008, 8m/1km  
Injection rate: 10 kg CO<sub>2</sub>/s for 20 years  
Monitoring period: 30 years

### Properties:

Typical of deep sandstone  
Stochastic distribution  
Hysteretic capillary and rel. perm functions

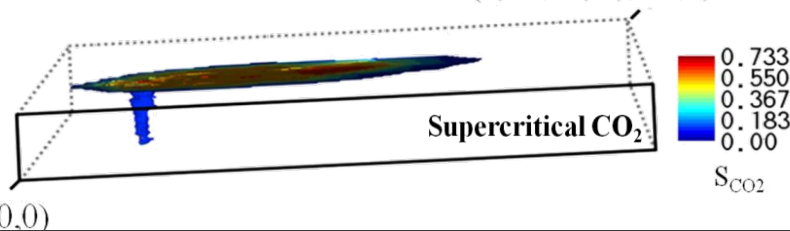
# Modeling results

## CO<sub>2</sub> injection as dissolved or supercritical

Supercritical

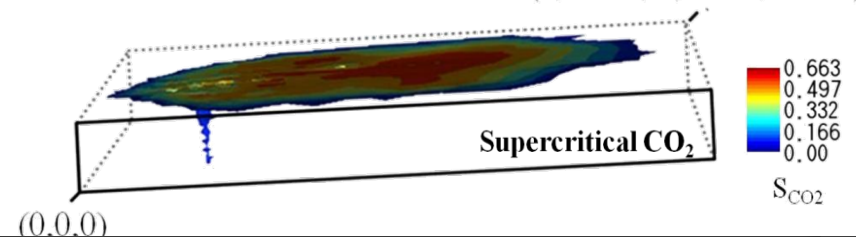
End of 20 Year Injection

(5,150m, 5,150m, 300m)



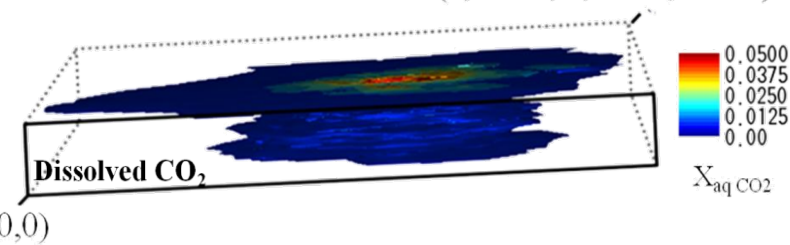
End of 30 Year Monitoring Period

(5,150m, 5,150m, 300m)

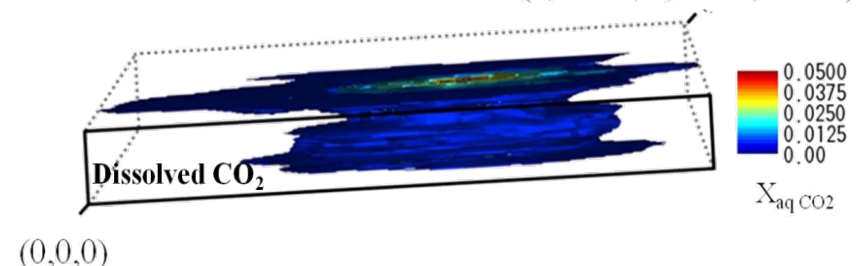


Dissolved

(5,150m, 5,150m, 300m)



(5,150m, 5,150m, 300m)



1 km

- Similar areal footprints after injection ~10 km<sup>2</sup>
- Supercritical CO<sub>2</sub> moves after injection, increasing area by 50% (14.9 km<sup>2</sup>)
- Dissolved CO<sub>2</sub> sinks after injection, decreasing area contacting caprock (8.9 km<sup>2</sup>)

# Conclusions

- Brine containing dissolved CO<sub>2</sub> can be mobilized upward by modest hydraulic gradients
- As the carbonated brine is depressurized, the CO<sub>2</sub> comes out of solution (exsolves) throughout the pore space
- The exsolved CO<sub>2</sub> phase has a very low relative permeability, even at high phase saturations. Exsolution relative permeability function
- Hysteric relative permeability represented by updating residual saturation in standard models. Simple, fits data well.
- Upward flow of brines containing dissolved CO<sub>2</sub> stops when the external driving force is removed, no runaway instability seen.
- Injection of CO<sub>2</sub> as a dissolved phase is likely to have a similar “footprint” to supercritical CO<sub>2</sub> injection, less mobile after injection.